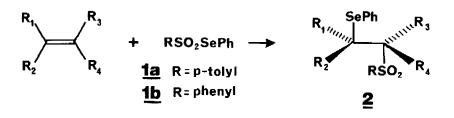
"SELENOSULFONATION" THE ADDITION OF SFLENOLSULFONATES TO OLEFINS Thomas G Back<sup>\*</sup> and Scott Collins Department of Chemistry The University of Calgary Calgary, Alberta, Canada, T2N 1N4

<u>Abstract</u> The reaction of phenyl areneselenolsulfonates with unhindered olefins affords  $\beta$ -phenylselenosulfone adducts, usually with high regio- and stereoselectivity

There is much current interest in the additions of selenenyl halides and pseudohalides to alkenes<sup>1</sup> Apart from being mechanistically intriguing, such reactions are of considerable synthetic value as the incorporation of selenium-containing residues into unsaturated systems permits their further functionalization by a variety of methods<sup>2</sup>. To our knowledge, selenolsulfonates  $\underline{1}$  have never been studied in this context

We have found that selenolsulfonates  $\underline{la}$  and  $\underline{lb}^3$  react with a variety of unhindered olefins to afford novel  $\beta$ -phenylselenosulfones  $\underline{2}$ , often in a highly regio- and stereoselective fashion. This process permits the simultaneous introduction of a phenylseleno group and a sulfonyl molety into the substrate, it is depicted below



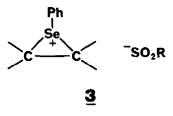
This "selenosulfonation" reaction is conveniently performed by treatment of the olefin with an equimolar amount of <u>la</u> or <u>lb</u> in dichloromethane solution at room temperature in the presence of a catalytic amount of boron trifluoride etherate The addition may also be effected in the absence of catalyst by refluxing the reactants in chloroform or benzene, or by heating them neat in a sealed glass tube The products were isolated by preparative tlc on silica-gel and are listed in the Table along with conditions and physical data

The boron trifluoride - catalyzed reaction of styrene with <u>la</u> provided the Markovnikoff adduct <u>2a</u> exclusively while similar treatment of methyl 10 - undecenoate with either <u>la</u> or <u>lb</u> gave <u>2c</u> and <u>2d</u> as *ca* 2.1 mixtures (nmr) of the two regionsomers which could not be separated. Markovnikoff addition provided the major product in both cases Furthermore, the addition of <u>la</u> to cyclohexene (catalyzed by boron trifluoride) and to indene (uncatalyzed) afforded solely the *trans* adducts <u>2e</u><sup>4</sup> and <u>2f</u><sup>5</sup> respectively

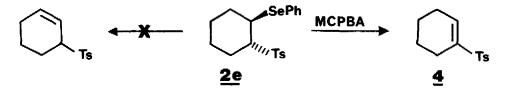
The uncatalyzed selenosulfonation of styrene and of allyl phenyl ether with <u>la</u> and <u>lb</u> respectively provided the anti-Markovnikoff adducts <u>2b</u> and <u>2g</u> in good yield The corresponding Markovnikoff products were not observed (tlc, nmr) Treatment of <u>2b</u> with a trace of boron trifluoride etherate in dichloromethane at room temperature failed to produce its regioisomer <u>2a</u>.

The addition of <u>la</u> to *trans*-5-decene was best effected in the absence of catalyst and solvent and provided a 9 l mixture (gc) of two stereoisomers. The major product was assumed to form *via anti* addition and was separated from its counterpart by crystallization. More hindered olefins failed to form adducts with <u>la</u> or <u>lb</u>

Clear conclusions regarding the mechanism of selenosulfonation await further study. However, the intermediacy of seleniranium ions <u>3</u> would account for the observed stereochemistry<sup>6</sup> At higher temperatures, product formation could also conceivably arise through a free-radical mechanism.



 $\beta$ -Phenylselenosulfones <u>2</u> cleanly undergo selenoxide elimination towards the activated hydrogen to afford vinyl sulfones<sup>7</sup>. For example, oxidation of <u>2e</u> with excess m-chloroperbenzoic acid in chloroform at room temperature gave an immediate exothermic reaction which furnished the known<sup>8</sup> sulfone <u>4</u> in 92% yield



The mechanism and synthetic utility of selenosulfonation as a method for elaborating olefins *via* the above and other procedures are under continued investigation in this laboratory.

## <u>Table</u>

## Selenosulfonation of Olefins

<u>Olefin</u>	Selenol- sulfonate	Product		Conditions <sup>C</sup> (time, h)	Isolated Yield (%)	m.p
Styrene	la	2a	$r_{s}$ $r_{s$	A (20)	86	159°
Styrene	la	2b	Ph PhSe Is	B (24)	93	157–158°
Methyl 10- undecenoate	la	2c	$\begin{array}{c} & \\ & \\ PhSe \\ 2 \end{array} \xrightarrow{R} + \\ Ts \\ SePh \\ 1 \end{array} \xrightarrow{R}$	A (24)	75	oil
Methyl 10- undecenoate	lb	2đ	$2^{R} + \frac{1}{PhSo_{2}SO_{2}Ph} + \frac{1}{PhSO_{2}SePh}$	A (24)	72	oıl
Cyclohexene	la	2e	SePh Junit Ts	A (18)	89	58-59°
Indene	1a	2f	O Jun SePh	B (72)	81	90-91°
Allyl phenyl ether	1b	2g	PhSO <sub>2</sub> SePh	C (67)	72	94-95°
<i>trans-5-</i> Decene	la	2h	$\begin{array}{c} n-C_{4}H_{9} \\ \end{array} \begin{array}{c} Ts \\ \\ FhSe \\ n-C_{4}H_{9} \end{array}$	D (21)	42	79-80°

<sup>(</sup>a) All products were identified by their <sup>1</sup>H and <sup>13</sup>C nmr and by their ir and mass spectra All solids gave correct combustion analyses, products 2c and 2d gave exact mass measurements in close agreement with calculated values (b)  $R = (CH_2)_{B}CO_2CH_3$  (c) A  $CH_2Cl_2$ , room temperature,  $BF_3 \cdot Et_2O$  catalyst. B CHCl<sub>3</sub>, reflux. C.  $C_6H_6$ , reflux D sealed glass tube, 75°.

## Acknowledgement

We thank Dr. R Yamdagnı for recording <sup>13</sup>C nmr and mass spectra Financial support from the Natural Sciences and Engineering Research Council and the Research Corporation is gratefully acknowledged

## References and Notes

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- 2 For reviews, see (a) H J Reich, <u>Acc Chem Res</u>, <u>12</u>, 22, (1979). (b) D.L.J. Clive, <u>Tetrahedron</u>, <u>34</u>, 1049, (1978)
- 3 Compounds <u>la</u> and <u>lb</u> were prepared from the corresponding sulfonhydrazides See T G. Back and S Collins, preceding paper
- 4 The <sup>1</sup>H nmr spectrum of <u>2e</u> showed signals at  $\delta$  (CDCl<sub>3</sub>) 3 14 (crude q, 1H) and 3 86 (crude q, 1H) Although the nmr spectrum does not permit unambiguous stereochemical assignment, the facile selenoxide elimination to form the vinyl sulfone <u>4</u> is most consistent with a *trans* diequatorial structure
- 5 The <sup>1</sup>H nmr spectrum of <u>2f</u> showed signals at δ (CDCl<sub>3</sub>) 3 2 (m, 2H), 3.8 (m, 1H) and 5 07 (d, J = 2Hz, 1H) Compound <u>2f</u> also readily underwent oxidation and elimination to afford the corresponding vinyl sulfone The *cis* structure would provide an unfavourable transition state for *sun* elimination in the observed direction
- 6 The addition of selenenyl halides and pseudohalides to olefins generally proceeds *via anti* addition For an interesting exception, see D.G. Garratt, <u>Can. J. Chem.</u>, <u>57</u>, 2180, (1979).
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(Received in USA 3 March 1980)